



How to Measure Metal Coating Thickness Using Handheld X-ray Fluorescence Analyzers

Handheld XRF is an indispensable tool in quality assurance that provides multiple benefits.

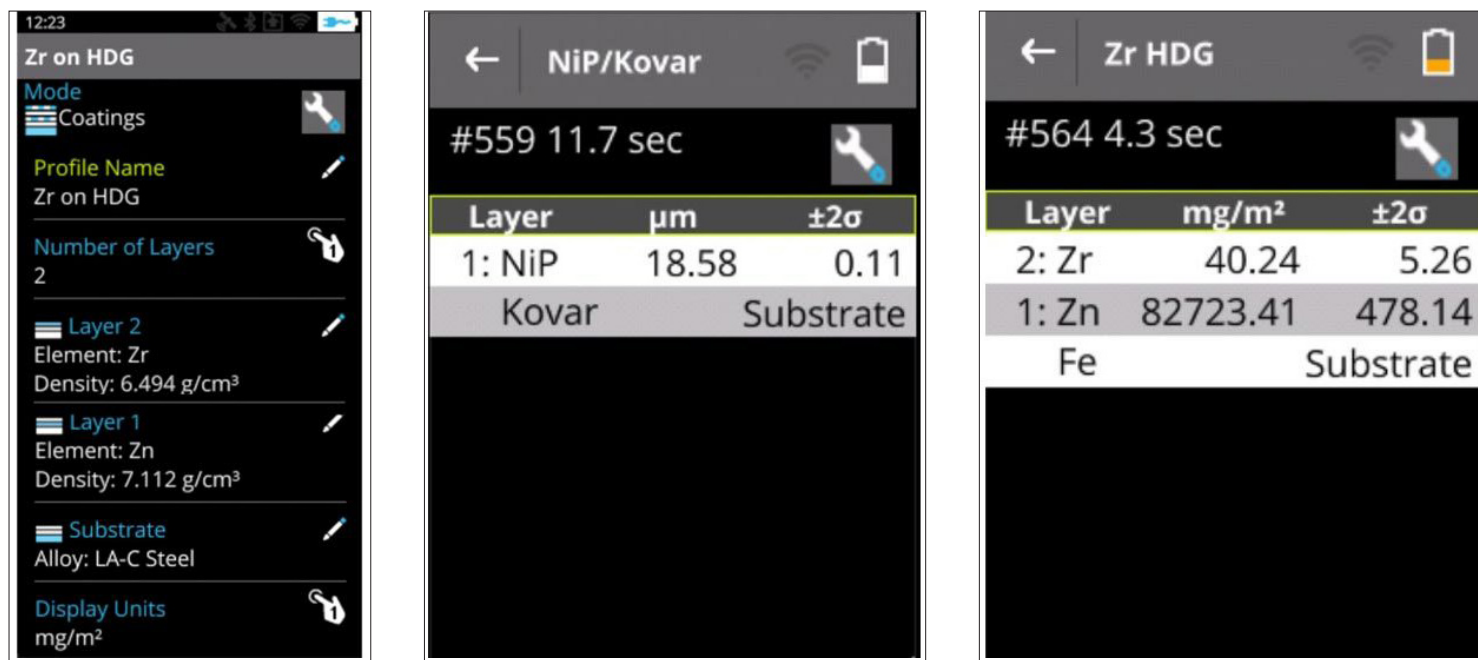
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Metal coatings are very commonly used in metal fabrication for decorative purposes or to enhance physical or chemical properties of metal made item surfaces. Metal coatings can be applied to enhance corrosion resistance, wear resistance, heat resistance, electrical conductivity, adhesion, solderability, and lubricity of metals. While over-coating significantly increases the cost of manufacturing, under-coating can cause product failure. To avoid these setbacks, controlling the coat weight or coating thickness is critical in metal finishing, fabrication, automotive, and aerospace industries to ensure components have the correct properties and optimize the cost of production at the same time.

NDT TECHNOLOGIES FOR COATING THICKNESS MEASUREMENTS

Numerous nondestructive testing technologies are available to measure the thickness of metal coatings at-line for quality control. These technologies are typically used at metal finishing companies or during the inspection process of raw material in industries using coated metals. While magnetic pull-off and magnetic induction gages measure non-magnetic coatings over iron and steel, eddy current gages measure the thickness of non-conductive coatings, generally paint, over non-magnetic substrates. In contrast, micro-resistance and ultrasound gages are suitable to measure the thickness of metal coatings over nonmetallic sub-

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A1 (left): Defining sample parameters. Entry of composition of substrate and coating layers. **A2** (center): Results of analysis for a sample with Zirconium over galvanized steel. Lab value for Zr: 38mg/m². Nominal Zn value: ca. 80 g/m². **B** (right): Results of analysis for a sample with 20µm NiP coating over Kovar (29Ni17Co, Balance Fe) substrate. *Image source Thermo Fisher Scientific.*

name stems from the user's ability to perform measurements without need of analytical calibration standards. However, it is recommended to use such standards to verify the accuracy of the instrument and apply corrections to improve accuracy as needed [2,3]. Nonetheless, the number of standards required is generally much lower than for empirical calibration models.

EXAMPLES

Although the fundamental parameter approach is versatile, it is required to know and define the composition of the substrate, the sequence, and composition of layers before performing the analysis. Data can be entered via the user interface of the instrument, as is shown in the example of figure A.1. In this first example, the substrate is made of a low alloy steel, the first layer is made of zinc and the second is a thin layer of zirconium. This type of galvanized steel coated with zirconium is used among others for car body panels. The layer of zinc prevents the oxidation of steel, while the layer containing zirconium further increases corrosion resistance and ensures better paint adhesion.

Next, one measurement was carried out without previously measuring any calibration standards. The results of analysis for the layers of zirconium (Zr: 40 mg/m²) and zinc (Zn:82723 mg/m²) match well respectively with the value determined using laboratory techniques for Zr (38 mg/m²) and the expected value for Zn (80000 mg/m²) as it can be seen in figure A2. The accuracy could possibly be improved further based on type standardization, by empirically adjusting the response of the analyzer to the known values of analytical calibration standards if deemed necessary.

Finally, another example of how fundamental parameter calibration can handle the analysis of a complex coatings system is outlined. The substrate is made of Kovar, an alloy that has a coefficient of expansion similar to glass which is used for glass sealing in microwaves, power or X-ray tubes. The Kovar alloy is coated with electroless nickel plating, an alloy made of nickel and phosphorous that provides substantial corrosion and wear protection. The system is complex for two reasons: the substrate and the coating are not simple metals but alloys and both also contain nickel

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at high concentrations. Despite this complexity, advanced fundamental parameter algorithm can work both in emission (the signal of the coating layer measured) and in absorption (the signal coming from the layer beneath or the substrate measured to calculate the attenuation caused by the layer above) and accurately determine the thickness of electroless nickel over Kovar (18.6 μm vs. 20 μm) as shown in figure B. Also, the results were obtained without previously measuring any calibration standards and may potentially be improved via type standardization.

CONCLUSION

Handheld XRF helps metal finishing businesses, and all other industries using metals, ensure metal coat weight or coating thickness specifications are met. Operators can perform analysis according to international methods such as ISO 3497 [2] and ASTM B568 [3]. Handheld XRF is an indispensable tool in quality assurance and quality control that provides multiple

benefits including:

- Verification of alloy grade and composition of incoming material, aside from coating
- Optimization of production costs and minimization of product failures
 - Coatings that are too thin can result in poor corrosion resistance
 - Coatings that are too thick lose money by using more product than necessary
 - Nondestructive analysis avoids cutting and damaging high-value product
- Improvement of quality by ensuring consistent coating across a product via multiple measurements
- Fast turnaround with real-time results and no sample preparation
- Generation of reports and certificates for quality assurance