

Introduction to FT-IR Sample Handling



Introduction

Why is it important to know about different methods of sample handling?

Certain techniques of sample handling are more effective than others for specific sample types. In order to obtain the best quality spectrum from your sample, it is important to know which handling technique works best for your sample type. Acquiring the best spectral data possible will give you more confidence in your results.

Sampling Techniques

Transmission

Source

How does it work?

The transmission technique does not require a separate accessory. The user simply places a sample directly into the infrared (IR) beam. As the IR beam passes through the sample, the transmitted energy is measured and a spectrum is generated. However, the analyst must often prepare the sample into a pellet, mull, film, etc. before the transmission measurement can be made. This requires expertise and can be time consuming.

Conceptual diagram of the beam path through a transmission sample

What types of samples can you analyze?

Excellent quality spectra can be obtained for many types of samples using transmission. The transmission technique can be used alone or in conjunction with accessories such as microscopes and liquid or gas cells to analyze:

- Organic powders in pellet or mull form
- Thermoplastic powders
- Soluble polymers
- Thin polymer films
- Regular-shaped polymers (with preparation)
- Irregular-shaped polymers (with preparation)
- Dark polymer films (not carbon-filled)
- Liquids (free-flowing or viscous)
- Gases
 (high concentrations to trace amounts)

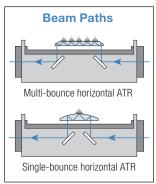
What are the advantages of transmission?

- Economical cells and mounts are generally inexpensive
- Well established most traditional form of sample measurement
- Excellent spectral information ideal for qualitative measurements
- Great for quantitative work many standard operating procedures are based on transmission

Attenuated Total Reflection (ATR)

How does it work?

An ATR accessory operates by measuring the changes that occur in an internally reflected IR beam when the beam comes into contact with a sample. An IR beam



is directed onto an optically dense crystal with a high refractive index at a certain angle. This internal reflectance creates an evanescent wave that extends beyond the surface of the crystal into the sample held in contact with the crystal. In regions of the IR spectrum where the sample absorbs energy, the evanescent wave will be attenuated. The attenuated beam returns to the crystal, then exits the opposite end of the crystal and is directed to the detector in the IR spectrometer. The detector records the attenuated IR beam as an interferogram signal, which can then be used to generate an IR spectrum.

What types of samples can you analyze?

ATR is ideal for strongly absorbing or thick samples which often produce intense peaks when measured by transmission. ATR works well for these samples because the intensity of the evanescent waves decays exponentially with distance from the surface of the ATR crystal, making the technique generally insensitive to sample thickness.

Other solids that are a good fit for ATR include homogeneous solid samples, the surface layer of a multi-layered solid or the coating on a solid. Even irregular-shaped, hard solids can be analyzed using a hard ATR crystal material such as diamond. Ideal solids include:

- Laminates
 Paints
 Plastics
 Rubbers
- Coatings
 Natural powders
 Solids that can be ground into powder

In addition, ATR is often the preferred method for liquid analysis because it simply requires a drop of liquid to be placed on the crystal. ATR can be used to analyze:

- Free-flowing aqueous solutions
 Coatings
- Viscous liquids
 Biological materials

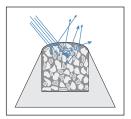
What are the advantages of ATR?

- Minimal sample preparation place the sample on the crystal and collect data
- Fast and easy cleanup simply remove the sample and clean the surface of the crystal
- Analysis of samples in their natural states no need to heat, press into pellets, or grind in order to collect spectra
- Excellent for thick or strongly absorbing samples ideal for difficult samples like black rubber

Diffuse Reflectance (DRIFTS)

How does it work?

When an IR beam is focused onto a fine particulate material, the incident beam can interact with the particle in one of several ways. First, radiation can be reflected off the top surface of the particle without penetrating the particle. Second, the light can undergo multiple reflections off particle surfaces without penetrating into the particle. True diffuse reflectance results from the penetration of the incident radiation into one or more sample particles and subsequent scatter from the sample matrix.



The IR beam interacting with a sample in a diffuse reflectance experiment

A DRIFTS accessory operates by directing the IR energy into a sample cup filled with a mixture of the sample and an IR transparent matrix (such as KBr). The IR radiation interacts with the particles and then reflects off their surfaces, causing the light to diffuse, or scatter, as it moves throughout the sample. The output mirror then directs this scattered energy to the detector in the spectrometer. The detector records the altered IR beam as an interferogram signal, which can then be used to generate a spectrum. Typically, a background is collected with the DRIFTS accessory in place and the cup filled with just the IR matrix. Excellent quantitative and qualitative data can be collected with proper sample preparation. However, transmission and ATR techniques are preferable to diffuse reflectance for quantitative data due to pathlength.

What types of samples can you analyze?

DRIFTS is commonly used for the analysis of both organic and inorganic samples that can be ground into a fine powder (less than 10 microns) and mixed in a powder matrix such as potassium bromide (KBr). Typical sample types include:

• Soft powders and powder mixtures • Hard polymers • Rigid polymers

The DRIFTS technique can also be used with silicon carbide paper for the analysis of large intractable surfaces. Silicon carbide paper can be used to rub off a small amount of a variety of samples for analysis. This technique is a viable alternative to traditional sampling techniques for:

Paint and varnish surfaces
 Tablets
 Rigid polymers

What are the advantages of diffuse reflectance?

- Little to no sample preparation just place in the sample cup
- Fast and easy cleanup dump the cup and blow or rinse clean
- No need for pressed KBr pellets or messy mulls samples can be run neat or diluted with KBr powder

True Specular Reflectance/Reflection-Absorption

How does it work?

True specular reflectance is a surface measurement technique that works on the principle of reflective efficiencies. This principle states that every sample has a refractive index that varies with the frequency of light to which it is exposed. Instead of examining the energy that passes through the sample, true specular reflectance measures the energy that is reflected off the surface of a sample, or its refractive index. By examining the frequency bands in which the rate of change in the refractive index is high, users can make assumptions regarding the absorbency of the sample. The true specular reflectance technique provides excellent qualitative data.

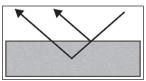


Diagram of the interaction of the beam using true specular reflectance

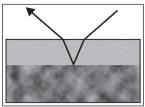


Diagram of the interaction of the beam using reflection-absorption

Reflection-absorption works on the same principle, but due to sample properties, some of the energy passes through the surface layer, is absorbed into the bulk of the sample, and then reflects off a substrate below the surface layer. A combination of true specular reflectance and reflection-absorption can occur when criteria for both techniques are met. If a qualitative comparison to transmission spectra is desired, users can apply the Kramers-Kronig correction to the data to remove the effects of dispersion.

What types of samples can you analyze?

Specular reflectance is commonly used for the analysis of both organic and inorganic samples having large, flat, reflective surfaces. Reflection-absorption can occur when one of the above criteria is compromised and the sample has a reflective substrate present just below the surface. This type of analysis is commonly used for:

Metallic surfaces

- · Silicon wafers
- Thin films on reflective substrates Laminated materials on metals

What are the advantages of specular reflectance?

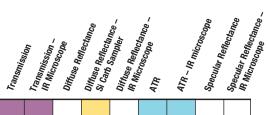
- Sensitivity to monolayer samples can detect Angstrom thick coatings on metal substrates
- Nondestructive analysis no contact or sample damage during analysis
- Wide range of accessories available can utilize main spectrometer and microscope accessorie

can utilize main spectrometer and microscope accessories depending on the size of the sample and the thickness of surface layer



* capable of being ground

† incapable of being ground



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Solids	Thermoplastic Polymers (can be melted)					
	Thermoplastic Polymers (can't be melted)					
	Soluble Polymers					
	Thin Polymer Films					
	Thick Polymer Films					
	Flat, Smooth Polymers *					
	Flat, Smooth Polymers [†]					
	Irregularly Shaped Polymers *					
	Irregularly Shaped Polymers [†]					
	Thin, Dark Polymer Films					
	Thick, Dark Polymer Films					
	Layered Polymer Films					
	Thin Polymer Film on Reflective Substrates					
	Thick Polymer Film on Reflective Substrates					
	Organic Powders					
	Adhesives					
	Rubber					
	Thin Fibers					
	Thick Fibers					
	Surface Analysis					
Liquids	Free-Flowing Aqueous Solutions					
	Other Free-Flowing Liquids					
	Viscous Liquids					
Gas	Gases (ppb to 100% concentration)					

Index of Sample Types

Powders – organic and inorganic solids that can be ground into a powder (2–5 micron particle size); Examples: chemicals, pharmaceuticals, crystalline materials, pigments, fibers, polymers and powders

Thermoplastic Polymers - polymers that can be pressed into free-standing thin films

Soluble Polymers – polymers that can be dissolved in a solvent or cast as a thin film

Thin Polymer Films – free-standing polymer films that are not thermoplastic or soluble and are less than 50 microns thick

Thick Polymer Films – free-standing polymer films that are not thermoplastic or soluble and are more than 50 microns thick

Regularly Shaped Polymers – polymers, films, and plaques that are hard or soft with a smooth surface, capable of being ground, not thermoplastic or soluble and regularly shaped

Regularly Shaped Polymers – polymers, films, and plaques that are hard or soft with a smooth surface, incapable of being ground, not thermoplastic or soluble and regularly shaped

Irregularly Shaped Polymers – polymers that are hard or soft with a rough or uneven surface, capable of being ground, not thermoplastic, or soluble and irregularly shaped; Examples: formed polymers, polymer beads and pellets

Irregularly Shaped Polymers – polymers that are hard or soft with a rough or uneven surface, incapable of being ground, not thermoplastic or soluble and irregularly shaped

Thin, Dark Polymers – carbon-filled polymers high in inorganic content that are not thermoplastic or soluble and less than 10 microns thick, such as carbon black

Thick, Dark Polymers – carbon-filled polymers high in inorganic content that are not thermoplastic or soluble and more than 10 microns thick

Layered Polymer Films – polymers that contain two or more layers or thin or thick films; Examples: layered paints and packaging materials

Thin Polymer Film on Reflective Substrate – polymer film on any kind of surface that reflects IR energy (usually metal) that is less than 15 microns thick; Examples: lubricants on hard disk media and layers on silicon wafers

Thick Polymer Film on Reflective Substrate – polymer film on any kind of surface that reflects IR energy (usually metal) that is more than 15 microns thick; Examples: coatings on containers (such as soda cans)

Adhesives - solid adhesives like tapes and solid glues

Rubbers – irregular-shaped rubber items that are not thermoplastic or soluble; Examples: o-rings, gaskets, and fittings

Thin Fibers - thin and bundled fibers

Thick Fibers - thick and bundled fibers

Surface Analysis - for qualitative analysis of the outermost layer of any solid or film

Free-Flowing Aqueous Solutions – liquids that contain any amount of water; Examples: inks, dyes, solvents, and paints

Other Free-Flowing Liquids - liquids that do not contain water

Viscous liquids - thick liquids, pastes, and emulsions; Examples: polyols, greases, and heavy oils

Gases (ppb to 100% concentration) – any sample that is a gas at room temperature or slightly above room temperature

Smart Accessories

Design elements to consider when choosing an accessory

We appreciate the need for efficiency and reproducibility in today's lab. We conducted extensive research regarding the features and benefits our users desire in their sampling accessories. This exhaustive process led to the development of unique Thermo Scientific[™] Smart Accessories[™]. These accessories offer the following value-added features:

- **Permanently aligned optics** needs no optical adjustment, so results are reproducible and quantifiable
- Rugged design protects optics from daily use and dust
- Automatic and fast purge achieves purge up to three times faster than a standard accessory
- Accessory recognition automatically identifies the accessory as soon as it is snapped in place and records its serial number in a non-editable history file
- **Experiment setup** automatically sets up your experiment parameters so you can start sampling immediately
- Accessory performance checks tests the accessory to ensure that it is performing optimally
- Spectral quality checks examines data as it is collected and rates the quality of the spectra you have collected, offering suggested improvements when necessary
- Multi-media tutorials and on-line help answers any questions you may have while conducting your analysis

We offer a comprehensive line of both Smart Accessories and standard sampling accessories to meet the needs of your laboratory. In addition, we offer a selection of microscope objectives to facilitate in the analysis of small samples.

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