## thermo scientific

## APPLICATION NOTE 43154

# Analysis of toxic elements in drinking and bottled waters using the Thermo Scientific iCAP 7200 ICP-OES Duo

## Authors

Sanja Asendorf, Application Specialist, Thermo Fisher Scientific, Bremen, Germany

## **Keywords**

Bottled water, Drinking water, Environmental analysis, Toxic elements

## Goal

This application note describes the analysis of several toxic elements in drinking and bottled waters using the Thermo Scientific iCAP 7200 ICP-OES Duo. The duo plasma enables excellent detection limits for toxic elements using axial view.

## Introduction

The increase in popularity of bottled drinking water has prompted many new regulations which manufacturers must adhere to. These apply to the country in which the water is sold and consumed. China and India have seen a huge increase in the consumption of bottled water in the last decades which has driven the contract analysis of toxic elements in these products to the following regulations in the respective countries (listed below with maximum limits expressed in Table 1).

## Chinese regulations:

- GB 8537-2008 Drinking natural mineral water
- GB 17324–2003 Hygienic standard of bottled purified water for drinking
- GB 5749–2006 Standards for drinking water quality
- GB 3838–2002 Environmental quality standard for surface water

## Indian regulations:

- IS 10500:2012 Drinking Water
- IS 13428:2005 Packaged natural mineral water
- IS 14543:2004 Packaged drinking water (other than packaged natural mineral water)



#### Table 1. Maximum permissible levels in mg·kg<sup>-1</sup>.

Element	GB 8537-2008	GB 17324-2003	GB 5749-2006	GB 3838-2002 (I) <sup>1</sup>	IS 10500:2012	IS 13428:2005	IS 14543:2004
Arsenic	0.01	0.01	0.01	0.05	0.01	0.05	0.05
Cadmium	0.003	-	0.005	0.001	0.003	0.003	0.01
Chromium*	0.05	-	0.05	0.01	0.05	0.05	0.05
Copper	1	0.01	1	0.01	0.05	1	0.05
Iron	-	-	0.3	0.3	0.3	-	0.1
Lead	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Mercury	0.001	-	0.001	0.00005	0.001	0.001	0.001
Nickel	0.02	-	0.02	0.02	0.02	0.02	0.02
Zinc	0.2	-	1	0.05	5	5	5

<sup>1</sup> For GB 3838, (I) refers to Class I categories, stated as mainly applicable to the source of water, National Nature Reserve.

\* For Chinese regulations, chromium is defined as hexavalent chromium present.

### Instrumentation

The Thermo Scientific<sup>™</sup> iCAP<sup>™</sup> 7200 ICP-OES Duo was used for the analysis. This is a compact dual view ICP-OES instrument based on the powerful core technologies of the Thermo Scientific iCAP 7000 Plus Series ICP-OES. The instrument achieves powerful analyte detection and provides a highly cost effective solution for routine analysis of liquids in laboratories with standard sample throughput requirements. The Thermo Scientific<sup>™</sup> Qtegra<sup>™</sup> Intelligent Scientific Data Solution<sup>™</sup> (ISDS) Software incorporates various functions for simplified method development and easy postanalysis data manipulation.

### Sample and standard preparation

A selection of drinking water samples (tap water and bottled water) were collected in China for analysis. In addition, European bottled water was also tested for comparison. The samples are listed below:

- Tap water sample from Dingpu river area, Shanghai
- Tap water sample from Jinqiao lake area, Shanghai
- Waterman (packaged drinking water)
- Nestle (natural mineral water)
- Evian (natural mineral water)

The samples did not require any pre-treatment and were analyzed directly after preservation in 0.5% analytical grade nitric acid ( $HNO_3$ ). Calibration standards were prepared in 0.5%  $HNO_3$  at the following concentrations: 0, 50 and 100 µg·kg<sup>-1</sup>. A QC check solution was prepared at 10 µg·kg<sup>-1</sup> to check recovery rate and test the stability of the method.

## Method development and analysis

A method was created with Qtegra ISDS Software and analytes added as indicated in Table 3. A standard sample introduction kit was used for the analysis as per the recommendations in the method notes. The method parameters are shown in Table 2.

#### Table 2. Methods parameters.

Parameter	Setting		
Pump Tubing	Sample Tygon <sup>®</sup> orange/white Drain Tygon <sup>®</sup> white/white		
Pump Speed	45 rpm		
Nebulizer	Glass concentric		
Nebulizer Gas Flow	0.19 MPa		
Spray Chamber	Glass cyclonic		
Auxiliary Gas Flow	0.5 L·min <sup>-1</sup>		
Coolant Gas Flow	12 L-min <sup>-1</sup>		
Center Tube	2 mm		
RF Power	1150 W		
Plasma View	Axial		
Exposure Time	5 s		

The samples were repeatedly analyzed in a single automated run over a period of 4 hours. Using the functionalities of Qtegra ISDS Software, a QC check was performed every 10 samples, recalibrating, recalculating and reacquiring from a previous sample whenever this check fails. A calibration was performed every 30 samples as per the requirements of the regulation.

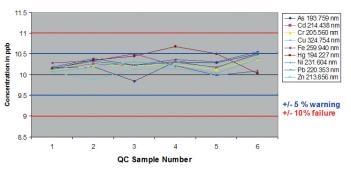
## **Results**

The samples were analyzed repeatedly in batches of 10 (2 of each of the 5 samples). Table 3 shows the averaged results of samples over the 4 hours and the method detection limits (MDLs). The concentrations found in all the samples were within the values outlined in the Chinese and Indian regulations. The MDLs are shown to be fit for purpose for this application. Nevertheless, the use of hydride generation accessories may be employed to further improve MDLs, particularly for mercury (to achieve sub µg·kg<sup>-1</sup> levels) when required.

Element and wavelength (nm)	MDL	Dingpu River	Jinquiao Lake	Waterman	Nestle	Evian
As 193.759	2.14	<dl< th=""><th>1.27</th><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	1.27	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Cd 214.438	0.07	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Cr 205.560	0.21	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Cu 324.754	0.39	<dl< th=""><th>1.52</th><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	1.52	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Fe 259.940	0.25	1.14	1.53	0.41	0.78	0.74
Hg 194.227	0.66	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Ni 231.604	0.36	1.05	0.57	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Pb 220.353	1.06	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>
Zn 213.856	0.19	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""><th><dl< th=""></dl<></th></dl<></th></dl<>	<dl< th=""><th><dl< th=""></dl<></th></dl<>	<dl< th=""></dl<>

Table 3. Averaged results and method detection limits in  $\mu g \cdot k g^{\text{-1}}.$ 

The 10 µg·kg<sup>-1</sup> QC check was used to check for recovery rates and drift during the run; which was found to be exceptionally stable as the chart in Figure 1 demonstrates. All QC recoveries were within 10% of their expected values throughout the 4 hour run.



#### QC stability at 10ppb over 4 hours

Figure 1. Stability of the 10  $\mu$ g kg<sup>-1</sup> QC check over 4 hours.

## Find out more at thermofisher.com/ICP-OES

For Research Use Only. Not for use in diagnostic procedures. ©2017 Thermo Fisher Scientific Inc. All rights reserved. Tygon is a trademark of Saint-Gobain Corporation. All other trademarks are the property of Thermo Fisher Scientific and its subsidiaries. This information is presented as an example of the capabilities of Thermo Fisher Scientific products. It is not intended to encourage use of these products in any manner that might infringe the intellectual property rights of others. Specifications, terms and pricing are subject to change. Not all products are available in all countries. Please consult your local sales representative for details. AN43154-EN 0717

## Conclusion

The analysis of environmental samples is rapid and analyst friendly using the Thermo Scientific iCAP 7200 ICP-OES Duo. The powerful and innovative design features of this instrument allow both novice and experienced analysts to quickly generate excellent results which allows a highly cost efficient sample analysis regime.

